

**Nomenclature**

Symbol	Quantity	Symbol
$a$	Acceleration	$m/s^2$
$a$	Specific Helmholtz function, $u - Ts$ ,	$kJ/kg$
$A$	Area	$m^2$
$A$	Helmholtz function, $U - TS$ ,	$kJ$
AF	Air-fuel ratio	
$c$	Speed of sound	$m/s$
$c$	Specific heat	$kJ/kg\cdot K$
$c_p$	Constant pressure specific heat	$kJ/kg\cdot K$
$c_v$	Constant volume specific heat	$kJ/kg\cdot K$
COP	Coefficient of performance	
$COP_{HP}$	Coefficient of performance of a heat pump	
$COP_R$	Coefficient of performance of a refrigerator	
$d, D$	Diameter	$m$
$e$	Specific total energy	$kJ/kg$
$E$	Total energy	$kJ$
EER	Energy efficiency rating	
$F$	Force	$N$
FA	Fuel-air ratio	
$g$	Gravitational acceleration,	$m/s^2$
$g$	Specific Gibbs function, $h - Ts$	$kJ/kg$
$G$	Total Gibbs function, $H - TS$	$kJ$
$h$	Convection heat transfer coefficient	$W/m^2\cdot ^\circ C$
$h$	Specific enthalpy, $u + Pv$	$kJ/kg$
$H$	Total enthalpy, $U + PV$	$kJ$
$h_c$	Enthalpy of combustion	$kJ/kmol$ fuel
$h_f$	Enthalpy of formation	$kJ/kmol$
$h_R$	Enthalpy of reaction	$kJ/kmol$
HHV	Higher heating value	$kJ/kg$ fuel
$i$	Specific irreversibility	$kJ/kg$
$I$	Total reversibility	$kJ$
$I$	Electric current	$A$
$k$	Specific heat ratio, $c_p / c_v$	
$k_s$	Spring constant	
$k_t$	Thermal conductivity	
$K_p$	Equilibrium constant	
ke	Specific kinetic energy, $V^2 / 2$	$kJ/kg$
KE	Total energy, $mV^2 / 2$	$kJ$
LHV	Lower heating value	$kJ/kg$ fuel
$m$	Mass	$kg$
$\dot{m}$	Mass flow	$kg/s$
$M$	Molar mass	$kg/kmol$
Ma	Mach number	
MEP	Mean effective pressure	$kPa$
$mf$	Mass fraction	
$n$	Polytropic exponent	
$N$	Number of moles	$kmol$
$P$	Pressure	$kPa$
$P_{cr}$	Critical pressure	$kPa$
$P_i$	Partial pressure	$kPa$

$P_m$	Mixture pressure	kPa
$P_r$	Relative pressure	
$P_R$	Reduced pressure	
$P_v$	Vapor pressure	kPa
$P_0$	Surroundings pressure	kPa
pe	Specific potential energy, $gz$	kJ/kg
PE	Total potential energy, $mgz$	kJ
$q$	Heat transfer per unit mass	kJ/kg
$Q$	Total heat transfer	kJ
$\dot{Q}$	Heat transfer rate	kW
$Q_H$	Heat transfer with high-temperature body	kJ
$Q_L$	Heat transfer with low-temperature body	kJ
$r$	Compression ratio	
$R$	Gas constant	kJ/kg·K
$r_c$	Cutoff ratio	
$r_p$	Pressure ratio	
$R_u$	Universal gas constant	kJ/kmol·K
$s$	Specific entropy	kJ/kg·K
$S$	Total entropy	kJ/K
$s_{gen}$	Specific entropy generation	kJ/kg·K
$S_{gen}$	Total entropy generation	kJ/K
SG	Specific weight or relative density	
$t$	time	s
$T$	Temperature	°C or K
T	Torque	N·m
$T_{cr}$	Critical temperature	K
$T_{db}$	Dry-bulb temperature	K
$T_f$	Bulk fluid temperature	K
$T_H$	Temperature of high-temperature body	K
$T_L$	Temperature of low-temperature body	K
$T_R$	Reduced temperature	
$T_{wb}$	Wet-bulb temperature	°C
$T_0$	Surroundings temperature	°C or K
$u$	Specific internal energy	kJ/kg
$U$	Total energy	kJ
$v$	Specific volume	m <sup>3</sup> /kg
$v_{cr}$	Critical specific volume	m <sup>3</sup> /kg
$v_R$	Pseudoreduced specific volume	
$V$	Total volume	m <sup>3</sup>
$\dot{V}$	Volume flow rate	m <sup>3</sup> /s
$V$	Voltage	V
$V$	Velocity	m/s
$V_{avg}$	Average velocity	
$w$	Work per unit mass	kJ/kg
$W$	Total work	kJ
$\dot{W}$	Power	kW
$W_m$	Work input	kJ

$W_{out}$	Work output	kJ
$W_{rev}$	Reversible work	kJ
$x$	Quality	
$x$	Specific exergy	kJ/kg
$x_{dest}$	Specific exergy destruction	kJ/kg
$X_{dest}$	Total exergy destruction	kJ
$\dot{X}_{dest}$	Rate of Total exergy destruction	kW
$y$	Mole fraction	
$z$	Elevation	m
$Z$	Compressibility factor	
$Z_h$	Enthalpy departure factor	
$Z_s$	Entropy departure factor	

**Greek letters**

$\alpha$	Absorbivity	
$\alpha$	Isothermal compressibility	1/kPa
$\beta$	Volume expansivity	1/K
$\Delta$	Finite change in quantity	
$\varepsilon$	Emissivity; effectiveness	
$\eta_{th}$	Thermal efficiency	
$\theta$	Total energy of a flowing fluid	kJ/kg
$\mu_{JT}$	Joule-Tomson coefficient	K/kPA
$\mu$	Chemical potential	kJ/kg
$\nu$	Stoichiometric coefficient	
$\rho$	Density	kg/m <sup>3</sup>
$\sigma$	Stefan-Boltzmann constant	
$\sigma_n$	Normal stress	N/m <sup>2</sup>
$\sigma_s$	Surface tension	N/m
$\phi$	Relative humidity	
$\phi$	Specific closed system exergy	kJ/kg
$\Phi$	Total closed system exergy	kJ
$\omega$	Specific or absolute humidity	kg H <sub>2</sub> O / kg dry air

**Subscripts**

<i>a</i>	Air
abs	Absolute
act	Actual
atm	Atmospheric
avg	Average
<i>c</i>	Combustion; cross section
cr	Critical point
CV	Control volume
<i>e</i>	Exit conditions
<i>f</i>	Saturated liquid
<i>fg</i>	Difference in property between saturated liquid and saturated vapor
<i>g</i>	Saturated vapor
gen	Generation
<i>H</i>	High temperature (as in $T_H$ and $Q_H$ )
<i>i</i>	inlet conditions
<i>i</i>	ith component
<i>L</i>	Low temperature (as in $T_L$ and $Q_L$ )
<i>m</i>	Mixture
<i>r</i>	Relative
<i>R</i>	Reduced
rev	Reversible
<i>s</i>	Isentropic
sat	Saturated
surr	Surroundings
sys	System
<i>v</i>	Water vapor
0	Dead state
1	Initial or inlet state
2	Final or exit state

**Superscripts**

· (over dot)	Quantity per unit time
-(over bar)	Quantity per unit mole
°(circle)	Standard reference state
*(asterisk)	Quantity at 1 atm pressure

**Some Physical Constants**

Universal gas constant	$R_u = 8.81447 \text{ kJ/kmol}\cdot\text{K}$
Standard acceleration of gravity	$g = 9.80665$
Standard atmospheric pressure	$1 \text{ atm} = 101.325 \text{ kPa}$
Stefan-Boltzmann constant	$\sigma = 5.6704 \times 10^{-8} \text{ W/m}^2\cdot\text{K}^4$
Boltzmann's constant	$k = 1.380650 \text{ J/K}$
Speed of light in vacuum	$c_0 = 2.9979 \times 10^8 \text{ m/s}$
Speed of sound in dry air at 0°C and 1 atm	$c = 2.9979 \times 10^8$
Heat of fusion of water at 1 atm	$h_{if} = 337.7 \text{ kJ/kg}$
Enthalpy of vaporization of water at 1 atm	$h_{fg} = 2265.5 \text{ kJ/kg}$